$\pm 0.2 \, {\rm deg \, kbar^{-1}}$ up to about 2.5 kbar. The Curie temperature at 1 atm was $-(21.9\pm0.1)^{\circ}$ C for the present specimen, which is in accord with $-(21.4\pm0.5)^{\circ}$ C reported by Yamada and Kamiyoshi.⁴⁾

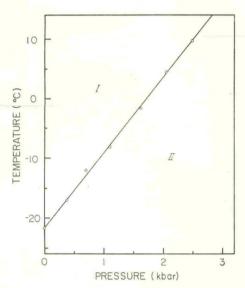


Fig. 4. Temperature dependence of the ferroelectric Curie temperature of RbDSO₄.

§ 4. Discussion and Conclusive Remark

The plus sign of the pressure coefficients of the Curie temperatures of RbHSO₄ and RbDSO₄ indicates that an order-disorder process of hydrogens (or deuteriums) in hydrogen bonds is not responsible to the ferroelectric phase transitions in these materials.¹²⁾ The above result is consistent with the peculiar isotope effect of RbHSO₄, and also with the results of recent deuterium quadrupole resonance study by Kasahara and Tatsuzaki¹⁸⁾ which showed that the deuteriums are in acentric positions even in the paraelectric state of RbDSO₄.

According to Chihara³⁾ the change in the specific heat at the Curie temperature of RbHSO₄ is 8.6 J deg⁻¹ mol⁻¹, then the Ehrenfest relation together with the present result of $dT_e/dp=12.0$ deg kbar⁻¹ and density of 2.89 g cm^{-8 2)} yields a change in the cubic thermal expansion coefficient at the Curie temperature as $\Delta\alpha=-6.2\times10^{-5}$ deg⁻¹. Yamada and Shimazu¹⁴⁾ showed that the changes in the linear thermal expansion coefficients along the b- and c-axis are -1.25×10^{-4} deg⁻¹ and 2.5×10^{-5} deg⁻¹, respectively. Then, the change in

the temperature coefficient of $(a \cdot \sin \beta)$ should be $3.8 \times 10^{-5} \, \text{deg}^{-1}$. The result indicates that the change in the linear thermal expansion of RbHSO₄ at the Curie temperature is rather uniaxial along the monoclinic axis of b.

The existence of the pressure induced Phase III of RbHSO₄ showed that the ferroelectric state of RbHSO4 is restricted in a limited region in the p-T plane alike to NH4HSO4. This would provide an important piece of information for those who intend to understand the ferroelectricity in these two isomophous bisulfates on a common basis. equilibrium II-III phase boundary was not determined because of very long time constant for the III-II transition especially at low temperatures. Thus, whether RbHSO4 has a lower Curie temperature at 1 atm in equilibrium seems still to be open to question; even if Phase III is stabilized at low temperatures at 1 atm, it would need very long time to realize the final equilibrium state.

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The authors would thank Professor I. Tatsuzaki of Hokkaido University for providing a single crystal of RbDSO₄. Thanks are also deboted for Dr. N. Yamada of Tohoku University and for Professor H. Chihara of Osaka University who showed experimental results prior to publication.

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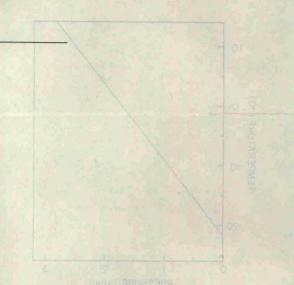


Fig. 4. Temperature dependence of the ferroelectric Curie temperature of RbDSO.

24. Disenssion and Conclusive Remark

The plus sign of the pressure coefficients of the Curie temperatures of RbHSO, and RbDSO, indicates that an order-disorder process of hydrogens (or deuteriums) in hydrogen bonds is not responsible to the ferroelectric phase transitions in those materials. The above result is consistent with the peopler isotope effect of RbHSO, and also with the results of recent deuterium quadrupole resonance study by Kasahara and Tatsurakian which showed that the deuteriums are in accentric positions even in the paraelectric

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The authors would thank Professor I. Tansuzaki of Hokkeldo University for providing a single crystal of RhDSO. Thanks are also deboted for Dr. N. Yamada of Tohoku University and for Professor H. Chihara of Osaka University who showed experimental results prior to oublication.

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